# **PSEUDO JAHN-TELLER ORIGIN OF THE PROTON-TRANSFER** ENERGY BARRIER IN THE HYDROGEN-BONDED [FHF] SYSTEM

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Abstract. The results of *ab initio* calculations of the adiabatic potential energy surfaces for the protonbound [FHF]<sup>-</sup> system at different F-F distances have been rationalized in the framework of the vibronic theory. It is shown that the instability of the symmetric  $D_{\infty h}$  structure at increased F...F distances and the proton displacement to one of the fluorine atoms are due to the pseudo Jahn-Teller mixing of the ground electronic state  $1^{1}\Sigma_{g}$  with the lowest excited state of  $\Sigma_{u}$  symmetry through the asymmetric  $\sigma_{u}$ vibrational mode.

Keywords: proton transfer, hydrogen bond, pseudo Jahn–Teller effect, potential energy surface, bifluoride anion.

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# Introduction

The proton transfer process underlies many important chemical reactions such as reductionoxidation interactions, high proton mobility in aqueous solutions, protons diffusion through membrane protein channels, etc. (see, for example, works [1-4] and references herein). A large number of experimental and theoretical works are devoted to the study of hydrogenbounded molecular systems [5-7]. The bifluoride anion plays an especially important role in the study of the hydrogen bond. It was established both experimentally and by the benchmark quantum-chemical calculations that the [FHF]<sup>-</sup> anion is linear and centrosymmetric in the gas phase, while in the crystal structures shifts of the proton from the mid-point are often observed [8-10]. The relationship between the F-H and F-F distances has been found and discussed [10]; in particular, it was shown that the longer the F-F distance, the greater the displacement of the H atom from the central position to one of the

Quantum-chemical calculations of such reaction [11-13].

fluorine atoms. systems are mainly devoted to the determination of equilibrium geometric parameters and to the calculation of the adiabatic potential energy surfaces (APES) for the elaboration of preparametrized analytical potential models that could be used to simulate the proton transfer

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On the other hand, it was shown earlier that the only reason of instability of the highsymmetry nuclear configurations and structural distortions of any molecular system in the nondegenerate state is the pseudo Jahn-Teller effect (PJTE), that is the vibronic mixing of considered state with the appropriate by symmetry excited states (see, for example, book [14] and reviews [15-17], and references therein). In systems with hydrogen bonds, a quantitative PJTE analysis was first used to study the origin of the hydrogen bond in the proton-bound ammonia dimer cation  $N_2H_7^+$  [18], and for a number of protonated proton-centered water clusters [19]. In these papers, it was shown that this approach reveals many details that will prove useful understanding hydrogen-bonded for systems.

The main goal of the present work is to demonstrate that with increasing F-F distances. the instability of the central position of a proton in the [FHF]<sup>-</sup> system and the appearance of a doublewell adiabatic potential energy surface (APES) for proton transfer are due to the pseudo Jahn-Teller effect.

## Method and computational details

The theory of the PJTE is well developed (see, e.g. [14,15]). In this approach the problem of the stability or instability of molecular nuclear configuration is reduced to the estimation of the curvature *K* of the adiabatic potential energy surface (APES) of the system in the arbitrary direction *Q* at the high-symmetry configuration  $Q_0$  for which the first derivatives are zero. In the second order perturbation theory with respect to small nuclear displacements *Q* the expression for *K* of any molecular system in its ground state  $|\Psi_0\rangle$ can be written as:

$$K = K_0 + K_v \tag{1}$$

The first term in Eq.(1),

$$K_0 = \langle \Psi_0 | (\partial^2 H / \partial Q^2)_0 | \Psi_0 \rangle \tag{2}$$

is the so-called primary force constant. It determines the restoring force arising when the nuclei are displaced with respect to the "frozen" electronic distribution. The second term, which is always negative,

$$K_{\nu} = -2\sum_{n} \frac{F_{on}^{2}}{E_{n} - E_{0}} = -2\sum_{n} \frac{\left| \left| \Psi_{0} \right| (\partial H / \partial Q)_{0} \left| \Psi_{n} \right| \right|^{2}}{E_{n} - E_{0}}$$
(3)

is the vibronic contribution to the curvature K of the adiabatic potential arising from mixing of the ground  $|\Psi_0\rangle$  and excited  $|\Psi_n\rangle$  electronic states. In the Eqs.(2) and (3) H is the adiabatic electronic Hamiltonian of the system which includes all Coulomb interactions between electrons and nuclei,  $|\Psi_i\rangle$  and  $E_i$  are, respectively, the electronic wave functions and the total energies of the ground and the excited states determined for the high-symmetry nuclear configuration, and  $F_{0n} = \langle \Psi_0 | (\partial H / \partial Q)_0 | \Psi_n \rangle$  in Eq.(3) are the constants of the vibronic coupling of the ground  $\Psi_0$  and the excited  $\Psi_n$  states. They are non-zero only if the product of irreducible representations of the wavefunctions  $\Psi_0$  and  $\Psi_n$  contains the irreducible representation of the displacement Q.

It was proved analytically and confirmed by a series of numerical calculations [14,15,20-22] that for any molecular system  $K_0 > 0$ . Hence, a negative curvature K and, therefore, the structural instabilities and distortions can be achieved only due to the negative vibronic coupling component  $K_{\nu}$ , and only if  $|K_{\nu}| > K_0$ . Thus, the value of  $K_0$ , the energy gaps  $\Delta_{0n} = E_n - E_0$  between the mixing states, and the vibronic coupling constants  $F_{0n}$  are the main parameters of the PJTE.

Direct calculation of the vibronic coupling matrix elements involved in the PJT models is rather difficult mathematically. In the present work the values of the PJTE parameters were estimated by fitting the *ab initio* calculated energy profiles (cross sections of the APES) of the [FHF]<sup>-</sup> system along the instability coordinate  $\sigma_{\rm u}$ (shifts of the proton from the mid-point of F...F distance) to the general formula, obtained from the vibronic theory. In the simplest case of the two-level PJTE problem when only one low-lying excited state contributes essentially to the instability of the ground state in the reference configuration, the roots of corresponding secular Eq.(4), where  $K_{01}$  and  $K_{02}$  are the primary force constants in the ground and the active excited state, respectively, and  $\Delta_{12}$  is the energy gap between the mixing states,  $\Delta_{12} = E_2 - E_1$ .

Note, however, that in any polyatomic system there are always many excited states of relevant symmetry, albeit at large energy gaps  $\Delta_{1n} = E_n - E_1$ , which mix to the ground state by vibronic coupling. Their contribution to the PJTE problem can be taken into account by the second-order perturbation corrections  $p_1$  and  $p_2$ , which can be given in Eq.(5) [23-25], where  $F_{1n}$ and  $F_{2n}$  are the respective vibronic coupling constants between the states  $|1\rangle$  and  $|2\rangle$  and the high-energy excited states  $|n\rangle$ . They may be rather small, just lowering the curvature of the mixing states from  $K_{01}$  and  $K_{02}$  to  $K_{01}$ - $p_1$ , and  $K_{02}$ - $p_2$ , and not producing instability of the ground state. With such overdetermined values of  $K_{01}$  and  $K_{02}$ , Eq.(4) for fitting the parameters takes the form in Eq.(6).

At small Q the resulting values of the curvatures of the APES of the two states  $K_1$  and  $K_2$  that are coupled by the PJT interaction of the Q direction, in Eq.(7).

$$\varepsilon_{1,2} = \frac{1}{4} (K_{01} + K_{02})Q^2 + \frac{\Delta}{2} \mp \frac{1}{2} \sqrt{\left[\frac{1}{2} (K_{01} - K_{02})Q^2 - \Delta\right]^2 + 4F^2 Q^2}$$
(4)

$$-p_1 = -2\sum_n \frac{|F_{1n}|^2}{E_n - E_1}$$
 and  $-p_2 = -2\sum_n \frac{|F_{2n}|^2}{E_n - E_2}$  (5)

$$\varepsilon_{1,2} = \frac{1}{4} (K_{01} - p_1 + K_{02} - p_2)Q^2 + \frac{\Delta}{2} \mp \frac{1}{2} \sqrt{\left[\frac{1}{2} (K_{01} - p_1 - K_{02} + p_2)Q^2 - \Delta_{12}\right]^2 + 4F_{12}^2 Q^2} \tag{6}$$

$$K_1 = (K_{01} - p_1) - 2F_{12}^2/\Delta$$
 and  $K_2 = (K_{02} - p_2) + 2F_{12}^2/\Delta$  (7)

The geometry optimization and electronic structure calculations of [FHF]- systems at different F…F distances in their proton-centered  $D_{\infty h}$  nuclear configurations were carried out in the frame of the RHF SCF method using the 6-31G\* basis set [26]. The potential energy curves along proton the off-center displacement were calculated taking into consideration configuration interaction (CI) with single and double excitations (CISD). The active space of CI included five occupied and two lower unoccupied molecular orbitals. All calculations were performed by using the PC GAMESS version [27] of the GAMESS (US) QC package [28].

#### **Results and discussion**

The investigation started with the [FHF]<sup>-</sup> molecule from the reference configuration of  $D_{\infty h}$ symmetry. The geometry optimization has shown that its equilibrium nuclear configuration corresponds to the symmetrical position of the H-atom, exactly in the middle of F…F distance (no imaginary frequency is observed). Calculated equilibrium F-F distance is equal to 2.24 Å. At the same time, the vibrational frequencies analysis of [FHF]<sup>-</sup> anions at increased F…F distances indicates the presence of one imaginary frequency (Table 1) corresponding to the low-symmetry  $\sigma_u$ -type distortion, which is mainly associated with the displacement of the proton towards one of the fluorine atoms. This means that  $D_{\infty h}$  configuration of these compounds is unstable with regard to the symmetrized  $\sigma_u$  displacements of its atoms. From Table 1 it can also be seen that with an increase in the F-F distance, the height and width of the barrier to proton transfer also increase.

The ground state wavefunction of the system belongs to the  ${}^{1}\Sigma_{g}$  representation of the  $D_{\infty h}$  symmetry point group; the valence electron configuration is ... $(3\sigma_{g})^{2}(1\pi_{u})^{4}(1\pi_{g})^{4}(3\sigma_{u})^{2}(4\sigma_{g})^{0}$   $(5\sigma_{g})^{0}(2\pi_{u})^{0}(6\sigma_{g})^{0}(2\pi_{g})^{0}$ . Some of these valence molecular orbitals are shown in Figure 1(*a*). According to the PJTE, excited states that produce the instability of the ground state should have the  ${}^{1}\Sigma_{u}$  symmetry:  ${}^{1}\Sigma_{g} \times {}^{1}\Sigma_{u} = {}^{1}\Sigma_{u}$ . The first excited state  $1{}^{1}\Sigma_{u}$  is formed by one-electron excitation from the HOMO  $3\sigma_{u}$  to the LUMO  $4\sigma_{g}$  (Figure 1(*a*)), the energy gaps  $\Delta_{01}$  between this state and the ground one,  $1{}^{1}\Sigma_{g}$ , varies in the range of  $10\div13$  eV, depending on the F-F distance.

Table 1

The values of imaginary frequencies  $\omega$  (cm<sup>-1</sup>), the height *h* (eV) and the width  $2Q_{\min}(\text{\AA})$  of the energy barrier.

	$R_{F-F}\left( \AA ight)$	$R_{F-F}(\AA)$ $\omega$		$Q_{min}$			
-	2.24	-	-	0.00			
	2.44	486.68	0.044	0.20			
	2.64	1333.84	0.340	0.35			
_	2.84	1523.78	0.759	0.50			

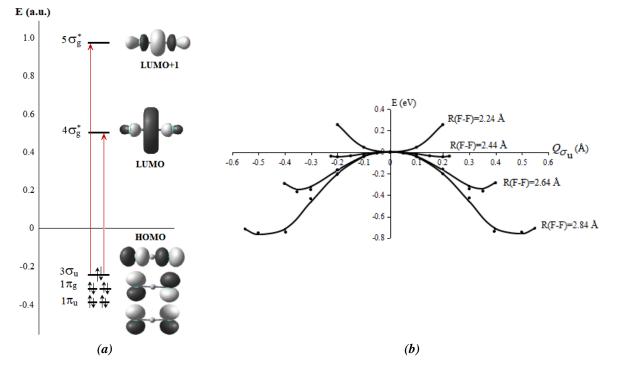


Figure 1. The MO energy level scheme for the [FHF]<sup>-</sup> molecules in the ground  ${}^{1}\Sigma_{g}$  state with indication of the one-electron excitations to the  ${}^{1}\Sigma_{u}$  terms (*a*); cross sections of the APES of [FHF]<sup>-</sup> along the  $\sigma_{u}$ -type distortion for several distances F...F (*b*); both *ab initio* and analytical (by Eq.(6)) calculations are shown by points and curves, respectively.

The second excited state,  $2^{1}\Sigma_{u}$ , with  $\Delta_{02}$ more than 20 eV is formed by excitation HOMO→LUMO+1. Subsequent higher-in-energy  ${}^{1}\Sigma_{u}$  excited states can be formed by one-electron excitations  $1\pi_g \rightarrow 2\pi_u$ ,  $1\pi_u \rightarrow 2\pi_g$ , etc. If one can assume that the main contribution to the instability of the ground state comes from only the first excited state  $1^{1}\Sigma_{u}$ , then the PJTE problem is simplified to a two-level one,  $(1^{1}\Sigma_{g} + 1^{1}\Sigma_{u})\otimes\sigma_{u}$ . The possible contribution of the second and all other higher lying excited  ${}^{1}\Sigma_{u}$  states to the curvature of the adiabatic potential of the ground state can be taken into account using the secondorder corrections  $-p_1$ , as described in the previous Section, Eq.(5). If the value of  $(K_{01}-p_1)$  in Eq.(6) is positive, this means that the higher excited states by themselves do not produce the instability of the ground state, and the lowest excited state  $1^{1}\Sigma_{u}$  is responsible for the main contribution to its instability.

To estimate the parameters of the vibronic coupling between the ground  $1^{1}\Sigma_{g}$  and the excited  $1^{1}\Sigma_{u}$  states, *ab initio* calculations of the energy profiles (APES cross sections) of the systems along the coordinate of instability  $\sigma_{u}$  were performed. Results for several distances F…F are shown in Figure 1(*b*).

It can be seen that at equilibrium  $R_{F-F}$  equals to 2.24 Å, the adiabatic potential is a single-well one, that is, the proton is located in the

middle between the fluorine atoms. Three lower curves in Figure 1(*b*) are the cuts of the APESes for  $R_{\text{F-F}}$ = 2.44 Å,  $R_{\text{F-F}}$ = 2.64 Å, and  $R_{\text{F-F}}$ = 2.84 Å, respectively. As the  $Q_{\sigma u}$  coordinate, it was used the proton displacement from the center of the F-F distance. It is seen that, with increasing F-F distances, the symmetric  $D_{\infty h}$  nuclear configuration becomes unstable with respect to the displacement of the central proton to one of the fluorine atoms, and the APESes become the double-well ones. The height and width of the energy barrier to the asymmetric displacement of the proton also increase.

By fitting Eq.(6) to the *ab initio* calculated energy profiles, we get the values of the parameter presented in Table PJTE 1. The parameters are as follows: the energy gaps  $\Delta$  between the ground  ${}^{1}\Sigma_{g}$  and the excited  ${}^{1}\Sigma_{u}$ terms, the primary force constants  $K_{01}$ - $p_1$  and  $K_{02}$ - $p_2$  in them, the vibronic coupling constants  $F_{12}$ , and the resulting values of the curvature of the ground state APES,  $K_{gr}$ , which at small Q can be approximately estimated according Eq.(7). It is important to note that the obtained value of  $K_{01}$ - $p_1$ from Table 2 is positive, showing that vibronic mixing of the ground state with higher-lying excited states does not produce the instability of the ground state, and it is the first excited state  $1^{1}\Sigma_{u}$  that is responsible for the main contribution to the instability.

Table 2

[FHF] <sup>-</sup> anions obtained by fitting Eq. (6) to the <i>ab initio</i> calculated energy profiles.								
$R_{F ext{-}F}\left( \AA ight)$	Δ	$F_{12}$	$K_{01}$ - $p_1$	$K_{02}$ - $p_2$	$K_{gr}$			
2.24	13.055	-	-	-	9.172			
2.44	12.796	16.193	36.541	52.259	-4.443			
2.64	11.826	15.980	30.847	32.796	-12.339			
2.84	10.570	13.956	21.782	21.848	-15.073			

PJTE coupling constants  $F_{12}$  (eV/Å), energy gaps  $\Delta$  (eV), primary force constants  $K_{01}$ - $p_1$  and  $K_{02}$ - $p_2$ , and resulting force constants  $K_{gr}$  (in eV/Å<sup>2</sup>) in the reference  $D_{\infty h}$  configuration of [EHE]: anions obtained by fitting Eq. (6) to the *ab initio* calculated energy profiles

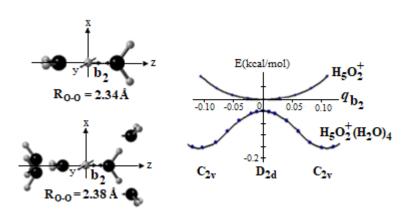


Figure 2. The cross sections of the APES along the off-center b<sub>2</sub>-type distortions for the Zundel cation  $H_5O_2^+$  and for the  $H_5O_2^+(H_2O)_4$  cluster.

It is also seen that the absolute value of the curvature of the adiabatic potential  $K_{gr}$  increases with increasing distance F...F, which results in an increase in the height and width of the energy barrier (Table 1).

A similar effect was obtained by us earlier for dihydronium cation  $H_5O_2^+$  [19]. As in the case of the [FHF]<sup>-</sup> anion, at the equilibrium O-O distance ( $R_{O-O}$ = 2.34 Å), the shared proton is located midway between the oxygen atoms (the single-well adiabatic potential in Figure 2). However, taking into account only the first solvation sphere around the Zundel cation in the  $H_5O_2^+(H_2O)_4$  cluster leads to a double-well adiabatic potential energy surface along the hydrogen bond coordinate and to the instability of the central position of the proton. The  $R_{O-O}$ distance in this case is equal to 2.38 Å (Figure 2).

# Conclusions

In this work, using the  $[FHF]^-$  system as an example, it has been shown that the hydrogen bonds in hydrogen bihalides  $[XHX]^-$  can be described in the framework of the pseudo Jahn-Teller effect; all calculated potential energy curves along the off-center proton displacements wholly coincide with those predicted from the general vibronic theory. Thus, the functional dependence of the potential energy on the instability coordinate  $\varepsilon(Q_{\sigma u})$  following from the PJTE theory, with the parameters estimated using quantum chemical calculations, can serve as a parametrized analytical model of the adiabatic potential, which can be used to simulate the proton transfer process in such systems.

Calculations confirm also that the breathing mode that controls the F-F distance between two fluorine atoms plays a very important role in the proton transfer dynamics in such systems. While at the equilibrium F-F distance the shared proton is located midway between both fluorine atoms, its central position becomes unstable with increasing F-F distance. This instability is shown to originate due to the pseudo Jahn-Teller mixing of the ground electronic state  $1^{1}\Sigma_{g}$  with the first excited state of  ${}^{1}\Sigma_{u}$  symmetry through the asymmetric  $\sigma_{u}$  vibrational mode.

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