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ADSORPTION OF STRONTIUM IONS FROM AQUEOUS SOLUTIONS ON NUT SHELLS ACTIVATED CARBONS

Gheorghe Duca, Mihai Ciobanu, Tudor Lupascu, Igor Povar *

Institute of Chemistry, 3, Academiei str., Chisinau MD 2028, Republic of Moldova
**e-mail: ipovar@yahoo.ca; phone: (+373) 22 73 97 36*

Abstract. The adsorption of strontium ions from aqueous solutions on nut shells activated carbons (samples CAN-7 and CAN-8) at different temperatures has been studied. The isotherm of adsorption of strontium ions from aqueous solutions on activated carbon CAN-7 has two inflection points at relatively small and high equilibrium concentrations. As the temperature increases, the adsorption values decrease, which indicates that the adsorption process is exothermic.

Keywords: activated carbon, adsorption heat, entropy, exothermic process, strontium ion.

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